

## A. Objective Questions

1. Write **true** or **false** for each statement :

- (a) The molecules of each substance are identical.
- (b) The inter-molecular forces are effective at all distances between the two molecules.
- (c) The molecules in a substance are in random motion.
- (d) In a gas, the molecules can move anywhere in space.
- (e) The liquids are less viscous than the gases.

**Ans.** (a) F (b) F (c) T (d) T (e) F

2. Fill in the blanks :

- (a) All the molecules of a substance are .....
- (b) The inter-molecular spacing is ..... in solids ..... in liquids and ..... in gases.
- (c) The molecular motion in liquid and gas is in ..... path.
- (d) In a solid, the molecules ..... but they remain at their fixed positions.
- (e) The inter-molecular forces are the weakest in .....
- (f) A solid exerts pressure .....
- (g) The gases are ..... dense.
- (h) A solid is ..... rigid.

**Ans.** (a) identical (b) least, more, still more  
(c) zig-zag (d) vibrate on either side  
(e) gases (f) downwards on its base  
(g) least (h) most

3. Select the correct alternative :

- (a) The diameter of a molecule is approximately
  - (i) 1 cm
  - (ii) 10 cm
  - (iii)  $10^{-10}$  m
  - (iv) 1 m
- (b) The inter-molecular forces are strongest in
  - (i) solids
  - (ii) liquids

(iii) gases

(iv) both (i) and (ii)

(c) The molecules

(i) in solid, liquid and gas, move freely anywhere.

(ii) in a solid, move freely within its boundary.

(iii) in a liquid, move within its boundary.

(iv) in a gas, move only within its boundary.

(d) The solids are

(i) more dense      (ii) less dense

(iii) least dense      (iv) highly compressible

(e) The inter-molecular forces in liquids are

(i) as strong as in solids

(ii) stronger than in solids

(iii) weaker than in solids

(iv) weaker than in gases

**Ans.** (a) (iii), (b) (i), (c) (iii), (d) (i), (e) (iii)

4. Match the following columns :

**Column A**

**Column B**

(a) A molecule is composed of      (i) does not exist free in nature.

(b) Ice, water and water vapour      (ii) can vibrate only up to about  $10^{-10}$  m from their mean positions.

(c) An atom      (iii) atoms.

(d) Gases      (iv) are the three states of water.

(e) The molecules of a solid      (v) occupy space

**Ans.** (a)-(iii), (b)-(iv), (c)-(i), (d)-(v), (e)-(ii)

**B. Short/Long answer questions**

1. Define matter. What is its composition ?

2. Name the *three* states of matter.

3. What is a molecule ?

4. What is a ...