Numericals : ALGI = ACI × ACG =

1. Convert 1 mm of Hg into pascal. Take density of Hg = 13.6×10^3 kg m⁻³ and g = 9.8 m s⁻².

Ans. 133.28 Pa

- 2. At a given place, a mercury barometer records a pressure of 0.70 m of Hg. What would be the height of water column if mercury in barometer is replaced by water? Take density of mercury to be 13.6×10^3 kg m⁻³.

 Ans. 9.52 m
 - 3. At sea level, the atmospheric pressure is 76 cm of Hg. If air pressure falls by 10 mm of Hg per 120 m of ascent, what is the height of a hill where the barometer reads 70 cm Hg. State the assumption made by you.

Ans. 720 m

Assumption: Atmospheric pressure falls linearly with ascent.

- 4. At sea level, the atmospheric pressure is 1.04×10^5 Pa. Assuming g = 10 m s⁻² and density of air to be uniform and equal to 1.3 kg m⁻³, find the height of the atmosphere.
 - Assuming the density of air to be 1.295 kg m⁻³, find the fall in barometric height in mm of Hg at a height of 107 m above the sea level. Take density of mercury = 13.6×10^3 kg m⁻³.

 Ans. 10 mm of Hg

Solution 1N.

1. Density of Hg = $13.6 \times 10^3 \text{ kgm}^{-3}$

Acceleration due to gravity, g = 9.8 ms⁻²

Height of mercury column = 1mm = 0.001 m

: Pressure, P = h,og

or,
$$P = (0.001)(13.6 \times 10^3)(9.8)$$
 pascal

Solution 2N.

2. Relative Density of Hg = $1.36 = 13.6 \times 10^{3} \text{ kgm}^{-3}$

Acceleration due to gravity, g = 9.8 m s⁻²

Height of mercury column = 0.70 m

: Pressure, P = h,og

or,
$$P = (0.7)(1.36 \times 10^3)(9.8)$$
 pascal

or,
$$P = 93.3 \times 10^3 \text{ Pa}$$

Let h be the height of water column

Then, P = h (density of water) g

or,
$$93.3 \times 10^3 = h \times 10^3 \times 9.8$$

or,
$$h = 9.52 \text{ m}$$

Solution 3N.

3. Atmospheric pressure, P = 76 cm Hg

Rate at which pressure falls = 10 mm of Hg per 120 m of ascent = 1 cm of Hg per 120 m of ascent

Let h be the height of the hill.

Pressure at hill, P' = 70 cm Hg

Total fall in pressure = P - P' = (76-70) cm Hg = 6 cm Hg

Now, fall in pressure is 1 cm Hg for every 120 m increase in height.

Thus, if the fall in pressure is 6 cm Hg, increase in height shall be (6×120) m = 720m

: Height of the hill = 720 m

Assumption: Atmospheric pressure falls linearly with ascent.

Solution 4N.

4. Atmospheric pressure, P = 1.04 × 105 Pa

Acceleration due to gravity, g = 10 ms-2

Density, $\rho = 1.3 \text{ kgm}^{-3}$

Let h be the height of the atmosphere.

$$P = h \rho g$$

$$h = \frac{P}{\rho g} = \frac{1.04 \times 10^5}{1.3 \times 10} = 8000 \text{ m}$$

Solution 5N.

Let h = 107 m be the height above sea level

$$\therefore P_h - P_{sea} = \rho_{air}gh$$

$$\therefore \rho_m g h_f - \rho_m g h_i = \rho_{air} g h$$

$$\therefore \rho_m g \Delta h = \rho_{air} g h$$

:
$$\Delta h = \frac{\rho_{air}h}{\rho_m} = \frac{1.295 \times 107}{13.6 \times 10^3}$$